100學年度研究所碩士班考試入學 生物技術研究所碩士班 生物化學考科

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單選題(共50題,每題2分)

- 1. What is the approximate p*K*a of a weak acid HA if a solution 0.1 M HA and 0.3 M A⁻ has a pH of 6.5? (A) 6.6 (B) 6.4 (C) 6.2 (D) 6.0 (E) 5.8
- 2. The blood buffering system is based on:
 - (A) the reaction of CO₂ with H₂O to form carbonic acid
 - (B) the ionization of aqueous carbonic acid to H^+ and the bicarbonate anion
 - (C) the decrease of the blood pH due to the production of H^+
 - (D) the excretion of bicarbonate and ammonium from the kidneys
 - (E) all of the above
- 3. The enzyme fumarase catalyzes the reversible hydration of fumaric acid to l-malate, but it will not catalyze the hydration of maleic acid, the cis isomer of fumaric acid. This is an example of:
 - (A) biological activity. (B) chiral activity. (C) racemization.
 - (D) stereoisomerization. (E) stereospecificity.
- 4. Which of the following statements about aromatic amino acids is correct?
 - (A) All are strongly hydrophobic.
 - (B) Histidine's ring structure results in its being categorized as aromatic or basic, depending on pH.
 - (C) On a molar basis, tryptophan absorbs more ultraviolet light than tyrosine.
 - (D) The major contribution to the characteristic absorption of light at 280 nm by proteins is the phenylalanine R group.
 - (E) The presence of a ring structure in its R group determines whether or not an amino acid is aromatic.
- 5. To determine the isoelectric point of a protein, first establish that a gel:
 - (A) contains a denaturing detergent that can distribute uniform negative charges over the protein's surface.
 - (B) exhibits a stable pH gradient when ampholytes become distributed in an electric field.
 - (C) is washed with an antibody specific to the protein of interest.
 - (D) neutralizes all ionic groups on a protein by titrating them with strong bases.
 - (E) relates the unknown protein to a series of protein markers with known molecular weights, $M_{\rm r}$.
- 6. Which of the following is an emerging technique for protein sequencing?
 - (A) NMR spectroscopy
 - (B) Mass spectrometry
 - (C) Gel electrophoresis
 - (D) Phylogenetic analysis
 - (E) Limited proteolysis

- 7. The residues tend to disrupt an α helix when they occur next to each other in a protein because: (A) an amino acids like The is highly hydrophobic.
 - (B) covalent interactions may occur between the Thr side chains.
 - (C) electrostatic repulsion occurs between the Thr side chains.
 - (D) steric hindrance occurs between the bulky Thr side chains.
 - (E) the R group of Thr can form a hydrogen bond.
- 8. Experiments on denaturation and renaturation after the reduction and reoxidation of the disulfide bonds in the enzyme ribonuclease (RNase) have shown that:
 - (A) folding of denatured RNase into the native, active conformation, requires the input of energy in the form of heat.
 - (B) native ribonuclease does not have a unique secondary and tertiary structure.
 - (C) the completely unfolded enzyme, with all disulfide bonds broken, is still enzymatically active.
 - (D) the enzyme, dissolved in water, is thermodynamically stable relative to the mixture of amino acids whose residues are contained in RNase.
 - (E) the primary sequence of RNase is sufficient to determine its specific secondary and tertiary structure.
- 9. The amino acid substitution of Val for Glu in Hemoglobin S results in aggregation of the protein because of ______ interactions between molecules.
- (A) covalent (B) disulfide (C) hydrogen bonding (D) hydrophobic (E) ionic 10. Which one of the following statements is true of enzyme catalysts?
 - (A) They bind to substrates, but are never covalently attached to substrate or product.
 - (B) They increase the equilibrium constant for a reaction, thus favoring product formation.
 - (C) They increase the stability of the product of a desired reaction by allowing ionizations, resonance, and isomerizations not normally available to substrates.
 - (D) They lower the activation energy for the conversion of substrate to product.
 - (E) To be effective they must be present at the same concentration as their substrates.
- 11. In competitive inhibition, an inhibitor:
 - (A) binds at several different sites on an enzyme.
 - (B) binds covalently to the enzyme.
 - (C) binds *only* to the ES complex.
 - (D) binds reversibly at the active site.
 - (E) lowers the characteristic V_{max} of the enzyme.
- 12. Which of the following is an epimeric pair?
 - (A) D-glucose and D-glucosamine (B) D-glucose and D-mannose
 - (C) D-glucose and L-glucose (D) D-lactose and D-sucrose (E) L-mannose and L-fructose

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- 13. A small molecule that *decreases* the activity of an enzyme by binding to a site other than the catalytic site is termed a(n):
 - (A) allosteric inhibitor. (B) alternative inhibitor. (C) competitive inhibitor.
 - (D) stereospecific agent. (E) transition-state analog.
- 14. A transition-state analog:
 - (A) is less stable when binding to an enzyme than the normal substrate.
 - (B) resembles the active site of general acid-base enzymes.
 - (C) resembles the transition-state structure of the normal enzyme-substrate complex.
 - (D) stabilizes the transition state for the normal enzyme-substrate complex.
 - (E) typically reacts more rapidly with an enzyme than the normal substrate.
- 15. Which of the following is not a reducing sugar?
 - (A) Fructose (B) Glucose (C) Glyceraldehyde (D) Ribose (E) Sucrose
- 16. In glycoproteins, the carbohydrate moiety is always attached through the amino acid residues:
 - (A) asparagine, serine, or threonine. (B) aspartate or glutamate.
 - (C) glutamine or arginine. (D) glycine, alanine, or aspartate.
 - (E) tryptophan, aspartate, or cysteine.
- 17. Which of the following best describes the cholesterol molecule?
 - (A) Amphipathic (B) Nonpolar, charged (C) Nonpolar, uncharged (D) Polar, charged(E) Polar, uncharged
- 18. An integral membrane protein can be extracted with:
 - (A) a buffer of alkaline or acid pH. (B) a chelating agent that removes divalent cations.
- (C) a solution containing detergent. (D) a solution of high ionic strength. (E) hot water.
- 19. Movement of water across membranes is facilitated by proteins called:
- (A) annexins. (B) aquaporins. (C) hydropermeases. (D) selectins. (E) transportins.20. Ubiquitin is a:
 - (A) component of the electron transport system. (B) protease. (C) protein kinase.
 - (D) protein phosphorylase. (E) protein that tags another protein for proteolysis.
- 21. Which of these cofactors participates directly in most of the oxidation-reduction reactions in the fermentation of glucose to lactate?
 - (A) ADP (B) ATP (C) FAD/FADH₂ (D) Glyceraldehyde 3-phosphate (E) NAD⁺/NADH
- 22. The anaerobic conversion of 1 mol of glucose to 2 mol of lactate by fermentation is accompanied by a net gain of: (A) 1 mol of ATP (B) 1 mol of NADH
 (C) 2 mol of ATP (D) 2 mol of NADH (E) none of the above.
- 23. Which of the following compounds *cannot* serve as the starting material for the synthesis of glucose via gluconeogenesis?

(A) acetate (B) glycerol (C) lactate (D) oxaloacetate (E) α -ketoglutarate

- 24. All of the following enzymes involved in the flow of carbon from glucose to lactate (glycolysis) are also involved in the reversal of this flow (gluconeogenesis) *except*:
 - (A) 3-phosphoglycerate kinase. (B) aldolase. (C) enolase. (D)phosphofructokinase-1.
 - (E) phosphoglucoisomerase.
- 25. The metabolic function of the pentose phosphate pathway is:
 - (A) act as a source of ADP biosynthesis.
 - (B) generate NADPH and pentoses for the biosynthesis of fatty acids and nucleic acids.
 - (C) participate in oxidation-reduction reactions during the formation of H₂O.
 - (D) provide intermediates for the citric acid cycle.
 - (E) synthesize phosphorus pentoxide.
- 26. Glycogen is converted to monosaccharide units by:
 - (A) glucokinase (B) glucose-6-phosphatase (C) glycogen phosphorylase
 - (D) glycogen synthase (E) glycogenase
- 27. Which of the following is not an intermediate of the citric acid cycle?
 - (A) Acetyl-coA (B) Citrate (C) Oxaloacetate (D) Succinyl-coA
 - (E) α-Ketoglutarate
- 28. Which of the following is not true of all naturally occurring DNA?
 - (A) Deoxyribose units are connected by 3',5'-phosphodiester bonds.
 - (B) The amount of A always equals the amount of T.
 - (C) The ratio A+T/G+C is constant for all natural DNAs.
 - (D) The two complementary strands are antiparallel.
 - (E) Two hydrogen bonds form between A and T.

29. In the laboratory, recombinant plasmids are commonly introduced into bacterial cells by:

- (A) electrophoresis a gentle low-voltage gradient draws the DNA into the cell.
- (B) infection with a bacteriophage that carries the plasmid.
- (C) microinjection.
- (D) mixing plasmids with an extract of broken cells.
- (E) transformation heat shock of the cells incubated with plasmid in the presence of CaCl₂.
- 30. Which one of the following analytical techniques does *not* help illuminate a gene's cellular function? (A) DNA microarray analysis (B) Protein chip analysis (C) Southern blotting (D) Two-dimensional gel electrophoresis (E) Two-hybrid analysis
- 31. Transamination from alanine to α -ketoglutarate requires the coenzyme:
 - (A) biotin. (B) NADH. (C) thiamine pyrophosphate (TPP).
 - (D) pyridoxal phosphate (PLP). (E) No coenzyme is involved.
- 32. Urea synthesis in mammals takes place primarily in tissues of the:
 - (A) brain. (B) kidney. (C) liver. (D)skeletal muscle. (E) small intestine.

33. The DNA binding motif for many prokaryotic regulatory proteins, such as the *lac* repressor, is: (A) helix-turn-helix. (B) homeobox. (C) homeodomain. (D) leucine zipper.

(E) zinc finger.

- 34. Which of the following are features of the wobble hypothesis?
 - (A) A naturally occurring tRNA exists in yeast that can read both arginine and lysine codons.
 - (B) A tRNA can recognize only one codon.
 - (C) Some tRNAs can recognize codons that specify two different amino acids.
 - (D) The "wobble" occurs only in the first base of the anticodon.
 - (E) The third base in a codon always forms a normal Watson-Crick base pair.
- 35. Processing of a primary mRNA transcript in a eukaryotic cell does not normally involve:
 - (A) attachment of a long poly(A) sequence at the 3' end.
 - (B) conversion of normal bases to modified bases, such as inosine and pseudouridine.
 - (C) excision of intervening sequences (introns).
 - (D) joining of exons.
 - (E) methylation of one or more guanine nucleotides at the 5' end.
- 36. An Okazaki fragment is a:
 - (A) fragment of DNA resulting from endonuclease action.
 - (B) fragment of RNA that is a subunit of the 30S ribosome.
 - (C) piece of DNA that is synthesized in the $3' \rightarrow 5'$ direction.
 - (D) segment of DNA that is an intermediate in the synthesis of the lagging strand.
 - (E) segment of mRNA synthesized by RNA polymerase.
- 37. Functional DNA is not found in:
 - (A) bacterial nucleoids. (B) chloroplasts. (C) lysosomes. (D) mitochondria.(E) nuclei.
- 38. Which of the following fatty acid is unsaturated?
 - (A) Arachidonic acid (B) Lauric acid (C) Palmitic acid (D) Stearic acid •
- 39. The molar concentration of pure water at 25°C is _____. (A) 36 M (B) 55.5 M (C) 10^{-14} M (D) 65 M °
- 40. An enzyme that transfers a phosphate group from ATP to a substrate is usually called _____.
 - (A) phosphatase (B) kinase (C) lipase (D) polymerase •
- 41. Which of the following amino acid *cannot* be phosphorylated?
 - (A) Tyrosine (B) Serine (C) Threonine (D) Glycine •
- 42. Which of the following enzyme can hydrolyze PIP₂ into IP₃ and DAG?
 - (A) Protein kinase C (B) Adenylate cyclase (C) Phospholipase C
 - (D) Glycogen synthase (E) Glycogen phosphorylase

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43. A major component of RNA but not of DNA is: (A) adenine. (B) cytosine. (C) guanine. (D) thymine. (E) uracil. 44. Which of the following methods does not separate molecules by size? (A) Gel filtration chromatography (B) Flow cytometry (C) Agarose gel (D) SDS-PAGE electrophoresis (E) Molecular sieve 45. DNA polymerase I in E. coli has proofreading activity, because it processes (A) $5' \rightarrow 3'$ exonuclease activity (B) $3' \rightarrow 5'$ exonuclease activity (C) $5' \rightarrow 3'$ endonuclease activity (D) $3' \rightarrow 5'$ endonuclease activity (E) None of the above 46. Which method apply polymerase chain reaction (PCR)? (A) Site-directed mutagenesis of a gene (B) ELISA (C) Northern blot (D) Yeast two-hybrid (E) Flow cytometry 47. Which of the following methods in not suitable to investigate protein-protein interaction? (B) Immunoprecipitation (A) Western blot (C) RNAi (D) Yeast two hybrid screening (E) Phage-display 48. Which description is *incorrect* about a cloning vector? (A) It usually carries an antibiotic resistance gene. (B) pET 23b(+) can accommodate insert up to 50 kb. (C) It usually carries a multiple cloning site. (D) It usually carries a replication origin. 49. Which of the following experiment can be used to study DNA-protein interactions? (A) RFLP (restriction fragment length polymorphism). (B) Gel Mobility Shift. (C) FISH (fluorescence in situ hybridization). (D) Yeast two-hybrid system. (E) Two-dimensional gel electrophoresis 50. The bacterium E. coli requires simple organic molecules for growth and energy-it is therefore a: (C) lithotroph. (A) chemoautotroph. (B) chemoheterotroph. (D) photoautotroph. (E) photoheterotroph.